

Synthesis of [8]Tritwistane† (Pentacyclo[8.4.0.0^{2,7}.0^{3,12}.0^{6,11}]tetradecane

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Summary The first synthesis of [8]tritwistane, which has D_3 symmetry, is reported.

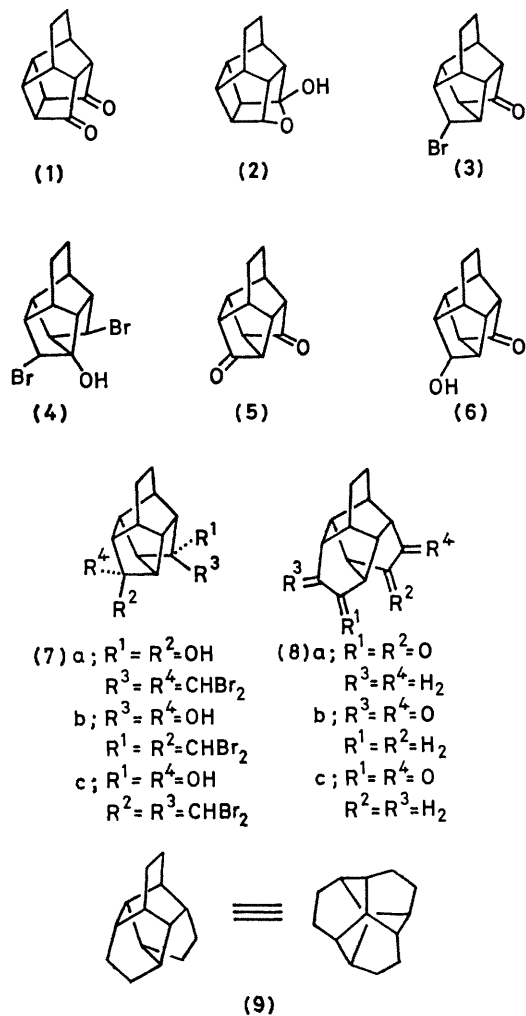
ALTHOUGH numerous methods for preparing adamantane derivatives have been studied, syntheses of the twistane series of compounds have hardly been developed. Previously a convenient method for the synthesis of ditwistane, a homologue of twistane, was established in our laboratory using catalytic hydrogenation as the key step.¹ [8]Tritwistane, the next highest homologue of ditwistane, is one of

the few compounds belonging to the chiral point group D_3 ,² and its synthesis is quite important for chiroptic studies.³ We have now synthesised [8]tritwistane from cyclohexa-1,3-diene in 34% overall yield.

The diketone (1), easily accessible from cyclohexa-1,3-diene-*p*-benzoquinone Diels–Alder adduct by photocyclization,⁴ was reduced with NaBH_4 to give the acetal (2)‡ (100%, m.p. 193–194 °C). Treatment of (2) with 47% HBr (110 °C, 13 h) produced the rearranged bromoketone (3) (87%, m.p. 83–86 °C) and the dibromoalcohol (4)

† Trivial name proposed by Schleyer *et al.*; cf. W. D. Graham, P. von R. Schleyer, E. W. Hagamane, and E. Wenkert, *J. Am. Chem. Soc.*, 1973, **95**, 5785.

‡ All new compounds gave satisfactory spectral and elemental analyses.



(10%, m.p. 121—122 °C). Oxidation of (3) with dimethyl sulphoxide in the presence of AgClO_4 gave the diketone (5) (66%, m.p. 195—197 °C) and the hydroxy ketone (6) (22%, m.p. 180—181 °C). Since (6) can be readily converted almost quantitatively into (5) by chromic acid oxidation, the conversion of (3) into (5) is quite efficient. The structural assignment of a symmetrical framework to (5) follows from its ^{13}C n.m.r. spectrum (CDCl_3): δ 17.55 (t), 31.59 (d), 32.73 (d), 41.00 (d), 45.44 (d), and 209.12 (s) p.p.m.

Expansion of the two cyclopentanone rings in (5) was stereospecifically achieved by the application of Nozaki's procedure.⁵ Thus, (5) reacted with CH_2Br_2 and lithium dicyclohexylamide in tetrahydrofuran (THF)-ether at -100 °C to give the tetrabromide (7) [83%, m.p. 279—281 °C (decomp.)], as sole product. This tetrabromide, on treatment with Bu^nLi in THF at -78 °C gave the desired diketone (8) [82%, m.p. 152—154 °C]. Compounds (7) and (8) must have the structures (7a) or (7b) and (8a) or (8b), respectively for the following reasons. The unsymmetrical (7c) and (8c) can be excluded easily by their ^1H n.m.r. spectra {(7) [$(\text{CD}_3)_2\text{SO}$]: δ 1.41 (br s, 4H), 1.84 (d, J 6 Hz, 2H), 2.04 (d, 6 Hz, 2H), 2.34 (m, 2H), 2.62 (m, 2H), 5.64 (s, 2H), and 6.28 (s, 2H); (8) (CDCl_3): δ 1.65 (m, 2H), 1.70 (s, 4H), and 2.1—2.7 (m, 10H)}. Compound (8) was converted into tritwistane (9) (76%, m.p. 100—102 °C) by desulphurization of the thioacetal of (8) (87%, m.p. 168—169 °C). The mass spectrum of (9) is characterized by a molecular ion peak at m/e 188 (100%) in addition to moderately intense peaks at m/e 173(14%), 161(23), 148(21), 131(21), 93(30), and 80(26). The ^1H and ^{13}C n.m.r. spectra of (9) in CDCl_3 attest D_3 symmetry of the molecule. At 270 MHz, there are three ^1H signals at δ 1.49 (m, 6H), 1.56 (m, 12H), and 1.75 (m, 2H). The ^{13}C -n.m.r. spectrum consists of only three peaks at δ 23.42 (t), 31.09 (d), and 35.96 (d) p.p.m.

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¹ K. Hirao, T. Iwakuma, M. Taniguchi, E. Abe, O. Yonemitsu, T. Date, and K. Kotera, *J. Chem. Soc., Chem. Commun.*, 1974, 691; K. Hirao, T. Iwakuma, M. Taniguchi, O. Yonemitsu, T. Date, and K. Kotera, *J. Chem. Soc., Perkin Trans. 1*, 1980, 163.

² Cf. M. Farina, *Tetrahedron Lett.*, 1963, 2097; G. R. Underwood and B. Ramamoorthy, *ibid.*, 1970, 4129; F. Vögtle and G. Hohner, *Angew. Chem., Int. Ed. Engl.*, 1975, 14, 497.

³ Cf. M. Farina and C. Morandi, *Tetrahedron*, 1974, 30, 1819.

⁴ R. C. Cookson, E. Crundwell, R. R. Hill, and J. Hudec, *J. Chem. Soc.*, 1964, 3062; D. Valentine, N. J. Turro, and G. S. Hammond, *J. Am. Chem. Soc.*, 1964, 86, 5202.

⁵ H. Taguchi, H. Yamamoto, and H. Nozaki, *J. Am. Chem. Soc.*, 1974, 96, 651.